

# SYNTHESIS AND CHARACTERIZATION OF TRIORGANOPHOSPHINEGOLD(I) THIOLATES

by
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## Additional References for Chapter 1.

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### Amended version of Table 4.4.2.

Complex	$H^2$	H <sup>9</sup>	H8	Phenyl Protons	$H_a$	$H_b$
6-MPH	8.45(s)	13.62(br,s)	8.26(s)	•	-	*
[dppm(Au(6-MP)) <sub>2</sub> ]	8.36(s)	13.12(br,s)	8.17(s)	7.95 - 7.37(br,m)	4.64(m)	•
[dppe(AuCl)(Au(6-MP))]	8.36(s)	13.25(br,s)	8.09(s)	7.87 - 7.47(br,m)	3.02(m)	
[dppe(Au(6-MP)) <sub>2</sub> ]	8.37(s)	13.13(br,s)	8.24(s)	7.89 - 7.49(br,m)	3.04(m)	*
[dppp(AuCl)(Au(6-MP))]	8.40(s)	13.22(br,s)	8.20(s)	7.74 - 7.50(br,m)	3.07(m)	1.73(m)
[dppp(Au(6-MP)) <sub>2</sub> ]	8.38(s)	13.18(br,s)	8.19(s)	7.83 - 7.42(br,m)	3.16(m)	1.89(m)

Note: Coupling constants, in parentheses, are in units of Hertz: a:  ${}^3J_{H-H}$ , b:  ${}^3J_{P-H}$  and c:  ${}^2J_{P-H}$ .

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**DECLARATION** 

This work contains no material which has been accepted for the award of any other degree or diploma in any university or other institution and, to the best of my knowledge and belief, contains no material previously published or written by another person, except where due reference has been made in the text.

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### ABSTRACT

The aim of this work was to synthesize a range of novel triorganophosphinegold(I) 6-mercaptopurinate complexes and to evaluate their potential anti-arthritic activity. The resultant complexes are based on the P-Au-S moiety, structurally related to the widely available pharmaceutical Auranofin, where the phosphorus atom is part of a triethylphosphine ligand and the sulphur atom derived from a tetraacetylated thioglucose anion.

Via alteration of the identity of the triorganophosphine group, a variety of complexes were synthesized from triorganophosphinegold(I) chloride precursors, falling into three main types: 1) triorganophosphinegold(I) 6-mercaptopurinate complexes with the general formula [R<sub>3</sub>PAu(6-MP)], where R<sub>3</sub>P = Et<sub>3</sub>P, Cycl<sub>3</sub>P, PhMe<sub>2</sub>P, Ph<sub>3</sub>P,  $(o-Tol)_3$ P,  $(m-Tol)_3$ P or (p-Tol)<sub>3</sub>P; 2) [μ-1,n-bis(diphenylphosphino)alkane]gold(I) chloride gold(I) 6-mercaptopurinate complexes with the general formula  $[(Ph_2P(CH_2)_nPPh_2)(AuCl)(Au(6-MP))]$  where n = 2 or 3; and 3) [µ-1,n-bis(diphenylphosphino)alkane]bis(gold(I) 6-mercaptopurinate) complexes with the general formula  $[(Ph_2P(CH_2)_nPPh_2)(Au(6-MP))_2]$  where n = 1, 2 or 3. These complexes and the triorganophosphinegold(I) chloride precursors were characterized using multinuclear magnetic resonance, infrared and Fast Atom Bombardment mass spectroscopic techniques. Unambiguous structure determinations of a selection of the complexes were achieved by single crystal X-ray crystallographic methods. Unit cell dimensions were: [PhMe<sub>2</sub>PAuCl], orthorhombic space group  $P2_12_12_1$ , a = 12.639(4), b = 16.931(6), c = 9.458(3) Å,  $V = 2024(1) \text{ Å}^3$  and Z = 4; [Ph<sub>3</sub>PAu(6-MP)].C<sub>2</sub>H<sub>5</sub>OH, triclinic space group  $P\bar{1}$ ,  $a=11.066(3),\ b=13.552(3),\ c=8.705(2)\ \text{Å},\ \alpha=91.51(2),\ \beta=113.06(2),\ \gamma=89.69(2)^{\circ},$  $V = 1200.8(5) \text{ Å}^3$  and Z = 2; and  $[(o\text{-Tol})_3\text{PAu}(6\text{-MP})].\text{C}_2\text{H}_5\text{OH}$ , monoclinic space group  $P2_1/n$ , a = 10.067(2), b = 10.518(2), c = 25.416(4) Å,  $\beta = 98.42(2)^\circ$ , V = 2662.1(9) Å<sup>3</sup> and Z = 4. The structures were refined to final R values of 0.035, 0.034 and 0.040, respectively, for reflections satisfying the  $I \ge 3.0\sigma(I)$  criterion: 1608, 3978 and 4183, respectively. The results for [PhMe2PAuCl] were utilized, in part, for a cone-angle to bond length correlation study on triorganophosphinegold(I) chloride complexes. The structures of [Ph<sub>3</sub>PAu(6-MP)] and [(o-Tol)<sub>3</sub>PAu(6-MP)] revealed a near linear P–Au–S chromophore, with angles of 173.71(6) and 177.03(8)°, respectively. A crystal structure analysis of a closely related triorganophosphinegold(I) thiolate complex, [Cycl<sub>3</sub>PAu(6p2-TU)], revealed a similar P–Au–S chromophore with an angle of 177.6(1)°. Unit cell dimensions were: monoclinic space group  $P2_1/c$ , a = 9.539(2), b = 16.452(4), c = 16.880(2) Å,  $\beta = 95.37(2)$ °, V = 2637.4(8) Å<sup>3</sup> and Z = 4. The final refinement value was R = 0.043, for 3695 reflections with  $I \ge 3.0\sigma(I)$ . The results for the three thiolate complexes were utilized in a correlation study of cone-angles to intramolecular parameters for triorganophosphinegold(I) thiolate complexes in general.

The combined microanalytical, spectroscopic and crystallographic studies verified the formation of all the thionucleobase complexes mentioned above and demonstrated that the gold centre is linearly bound to both the phosphorus and sulphur atoms.

A number of the triorganophosphinegold(I) 6-mercaptopurinate complexes were tested for their anti-arthritic activity in Dark Agouti rats with promising results.

### **ABBREVIATIONS**

Å Angström

br broad

°C degree Celsius

13C NMR carbon-13 nuclear magnetic resonance

Cycl cyclohexyl

d doublet

dd, dm, dt doublet of doublets, multiplets, triplets

dec. decomposition point

dppe  $\mu$ -1,2-bis(diphenylphosphino)ethane

dppm μ-bis(diphenylphosphino)methane

dppp  $\mu$ -1,3-bis(diphenylphosphino)propane

Et ethyl

FAB-MS Fast Atom Bombardment - mass spectroscopy

g gram

<sup>1</sup>H NMR proton nuclear magnetic resonance

Hz hertz

IR, ir infrared spectroscopy

xJ<sub>AB</sub> coupling constant between nuclei A and B over x bonds

m multiplet (nmr)

m medium (ir)

M molar (mol dm<sup>-3</sup>)

[M]<sup>+</sup> molecular ion

2mbaH 2-mercpatobenzoic acid

6m2-TUH 6-methyl-2-thiouracil

Me methyl

MHz mega-hertz

ml

millilitre

mmol

millimole

m.p.

melting point

6-MPH

6-mercaptopurine

MW

molecular weight

m/z

mass to charge ratio

6p2-TUH

6-n-propyl-2-thiouracil

NMR, nmr

nuclear magnetic resonance (Fourier Transform)

N.O.

not observed

Obs.

obscured

31P NMR

phosphorus-31 nuclear magnetic resonance

Ph

phenyl

PhO

phenoxy

ppm

parts per million

q

quartet

S

singlet (nmr)

S

strong (ir)

sh

shoulder

t

triplet

**TMS** 

tetramethylsilane

m-Tol `

meta-tolyl

o-Tol

ortho-tolyl

p-Tol

para-tolyl

2-TUH

2-thiouracil

vs

very strong

w

weak

%Y

percentage yield

 $\delta(A-B)$ 

bending frequency of A-B bond

λ

wavelength

 $\nu(A-B)$ 

stretching frequency of A-B bond

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