

The second of the series of scientific lectures undertaken by Professor Rennie was delivered at the University on Thursday evening, when the constituents of the atmosphere were demonstrated in their various uses and functions. The movements of the particles of gases were first dealt with, the lecturer demonstrating the varying speed with which particles in their various states pass through one another under obstructing elements. It was shown that motion, more or less rapid, was a necessary condition to the existence of atmosphere, whether the particles were gaseous, liquid, or solid. By way of illustrating this it was stated that if a vessel containing air were inverted into a room previously emptied of air would rapidly diffuse itself into every portion of the room. It was then shown by a number of experiments how this diffusion was accompanied by gases varying in weight, some moving more rapidly than others. The reason of the atmosphere remaining upon the earth's surface was shown to be because the force of gravity overcame the rate of speed at which the gases moved. That there was no atmosphere above the surface of the earth was due to there being no sufficient gravity to retain the oxygen and nitrogen. The differing theories regarding the surface of the planet Mars were discussed, the question at issue being as to whether the water supposed to exist there was in a liquid or vaporous state. The lecture concluded with a number of interesting experiments showing the solution of gases in water and other liquids, and the effect of such solutions. Were it not for the fact that oxygen was soluble in water animal life could have no existence. It was, however, shown that the experiments set down upon the syllabus for the evening, the Professor demonstrated that gases had a great tendency to adhere to the surfaces of solids. In the case of platinum it was shown that this metal retained oxygen more tenaciously than any other element except itself. The experiments were very closely scrutinized, and the lecturer retained the keen attention of his listeners.

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UNIVERSITY EXTENSION LECTURES.

Professor Rennie continued his series of lectures on "The Atmosphere" before a large audience at the University on Thursday evening. Having enumerated the principal constituents of the air the lecturer proceeded to describe in detail the most important of them, namely oxygen. He gave an account of its properties, and particularly, both on the small and the large scale, and illustrated its action by experiments. He then gave some account of the part of oxygen played in respiration and the purification of the blood. In this connection he described the various uses to which it can be applied. Ozone was then mentioned, and its preparation and properties illustrated by experiments. The lecturer showed that although doubtless ozone acted as a purifying agent upon the atmosphere, yet considerable doubt had of late been thrown upon the idea that ozone possessed germ-killing properties.

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30. UNIVERSITY EXTENSION LECTURES.

Professor Beale continued his course of lectures on "Rome" before large audiences, at the University on Wednesday afternoon and evening. The preceding lecture dealt with the building and architectural style of architecture as an introduction to a description of the Forum Romanum, which was the most instructive and interesting centre in the city of Rome and all its historical associations. This portion of the city had always made the most interesting, because the Forum was the scene of all the functions which took place in the city, whether of legal, political, religious, military, commercial, or social character. The lecturer went on to describe these functions as though he were a Ciceronian reporter from the life of the ancient days, and knew everything that went on to the minutest details. In fact, one would think that the modern guides who make a living by conjuring with the ignorance of sightseers to the glorious ruins of the great city would not dare to approach the subject so closely as did the Professor. His first illustration was a map showing the position of the Forum, and the intelligence of the spectators was aided by a graphic description of the battle between the Sabines and the Romans on the issue of which was decided by Romulus's stratagem. At first the Forum was a market-place, with booths and shops scattered about promiscuously for the transaction of commercial pursuits, but the King Tarquinus Urius laid it out in an organized arrangement for shops around a large square for the assembly of the people, armaments and other mass meetings of general interest. The public buildings, such as temples dedicated for various purposes, a prison—in which the Apostle Peter is supposed to have been kept—rostra, upon which the orators and orators delivered their soul-stirring speeches, and the Rostra. Halls were erected in various suitable places within the Forum area. The temples, their origin and uses, claimed the special attention of the Professor, his description of them in their original beauty and grandeur being of great interest to the students of architectural design and archaeological research. After showing the Forum in all its most fascinating picturesqueness, and thus playing upon the auditory imagination of the listeners, the lecturer added some notices of what now

remained of the greatest productions of Roman genius, aided by skill and art, in most luxuriant profusion. The lecturer, however, gave a certain impression that to realize what had just been described needed all the enthusiasm that none but such as the Professor himself could possibly entertain. The authentic stories and the legends which were given to explain the names and titles of the various buildings and their associations kept the audience intensely interested during the whole of the lecture.

The third of Dr. Rennie's lectures upon the atmosphere was delivered at the University on Thursday evening before a good attendance of students and interested spectators. The lecturer proceeded to explain the composition of the atmosphere, showing that although the air contained a number of gases in greater or less quantities, the two principle proportions were those of oxygen and nitrogen, and these had of late to be altered to admit of the newly discovered element of argon. Oxygen being the most important of gases, Oxygen being the main topic for the discourse. The discovery of oxygen as a diatomic gas, on August 1, 1774, by Dr. Priestly, was referred to as one of the most important discoveries in science. The operation of the electric spark may also be mentioned, the application of heat to red oxide of mercury was the subject of an interesting experiment. Pure oxygen having been obtained, it was clearly demonstrated that this constituent in the air was the necessary element for combustion, a piece of metal having been placed into a jar of oxygen and resulted in the metal being rapidly burned, nothing of it being left but the rusty film attaching to the sides of the glass. Zinc and even platinum could be easily burnt in the same way. The lecturer then went on to treat of the practical uses to which oxygen could be put. In this connection, the artificial preparation in large quantities being obtained by heating a number of substances; chlorate of potash was the most generally used, although oxide of barium and sodium peroxide were well known. The relation of oxygen to animal life was next treated, the Professor explaining the diffusion of the gas upon the lungs and the blood to sustain life. Various experiments upon animal life had been made showing that pure air in the form of oxygen was continually breathed life would remain in a very short space of time, which demonstrated that pure air was usually breathed was quite good enough for ordinary purposes of living. It had, however, come into the practice of medicine as a stimulant, and was said to have been used to prolong the life of the late Czar of Russia. It was also used in the manufacture of spirits, in the manufacture of sugar, bleaching, and manufactures of various kinds. The lecture was concluded with an interesting experiment, showing the remarkable conversion of oxygen into ozone by contact with electricity, and explaining the properties of ozone. The lecture incidentally remarked that although ozone existed as a gas under certain atmospheric conditions, the decomposition of seaweed was not ozone, as popularly believed, although its odour was rather like it. At the next lecture Professor Rennie will show the students demonstrating the influence exerted by nitrogen upon the peculiar forms of growth in plant life. These experiments are new, and have never been shown here before. They will be illustrated by lantern slides and microscopically.